SECTION

Synonyms C.A.S. No. Unit(s) Size Formula Product SECTION II 100, 500 grams, 2.5 Kg. SODIUM CARBONATE, MONOHYDRATE 5968-11-6 Na 2CO3·H2O Sodium Carbonate

հեր 716-226-6177 Napri 716-334-4222 800-424-9300 CHEMTREC Reactivity L Health Fire 0

DEPT PART **HAZARD RATING** MOUNT HATE

3W HIX

HAZARDOUS INGREDIENTS OF MIXTURES

Emergency and First Aid Procedures

Principal Hazardous Component(s) CAUTION! CAUSES SKIN AND EYE IRRITATION-HARNFUL IF Sodium Carbonate, Monohydrate

100% 8

None established

TLV Units

SECTION III PHYSICAL DATA

Boiling Point (°F) Vapor Pressure (mm Hg) | Negligible as solid 851°C (1564°F) Decomposes, begins at

Melting Point (°F)

752°F $\frac{E_{\text{NBBR}}}{N} = \frac{1}{N} \frac{|N_{\text{OB-VOlatile}}(NA)}{N}$ Percent Volatile by Volume (%) Specific Gravity (H,0=1) 1.55 at 25°C Non-volatile (NA).

Vapor Density (Air = 1) Data not listed

33 grams per 100 ml at 20°C.

Solubility in Water

Appearance and Odor |White crystalline powder or granular; no odor SECTION IV

FIRE AND EXPLOSION HAZARD

% by Volume

DATA * * * * *

Upper

Extinguisher Media (Method Used) Non-combustible (NA).

Mone required. Product inherent fire retardant.

SPECIAL FIREFIGHTING PROCEDURES

Its presence in a fire does not hinder the use of any standard extinguishing media. Wear a NIOSH-approved self-contained breathing apparatus.

UNUSUAL FIRE AND EXPLOSION HAZARDS

None. Note decomposition beginning temperature, Section III and decomposition products, Section VI. Can react vigorous! 20 25 26 27 28 293031/ react_vigorously

NON-REGULATED

D.O.T.

FREY SCIENTIFIC

SCIENTIFIC

HICKORY LANE MANSFIELD OHIO 44905

Sold Sales S C. 1811-11

> Threshold Limited Value SECTION V HEALTH HAZARD DATA

Effects of Overexposure

Mone established.

LD₅₀ 2.8 g/Kg. (16 CFR 1500.3).

<u>IHMAATION:</u> High dust concentrations may cause upper respiratory tract irritation. <u>ETES:</u> Contact with the powder may cause eye irritation. SKIN: Prolonged or repeated skin contact may cause skin irritation. SMCESTION: Of large quantities may occur and produce corrosion of Contact with the powder may cause eye irritation.

G. I. tract, vomiting, diarrhea, circulatory collapse, death

immediately flush eyes with plenty of water and get medical attention.

MGESTION: If swallowed, do NOT induce vomiting. If conscious, give one

or two glasses of water and call physician.

SECTION VI REACTIVITY DATA

Stability Unstable Stable Conditions to Avoid Excessive temperature and heat. Acids and acid fumes

Aluminum, Fluorine, Phosphorus Pentoxide. Acids cause decomposition liberating gaseous carbon dioxide

Decomposition Products Carbon dioxide. (CO2)

Hazardous

(Materials to avoid) Incompatibility

Hazardous Polymerization Conditions to Avoid

May Occur SECTION VII SPILL OR Will Not Occur LEAK PROCEDURES

Sweep up and place in a suitable container

Steps to be taken in case material is released or spilled

Waste Disposal Method

Dissolve in water and flush to sewer with copious amounts of water

Discharge, trealment, or disposal may be subject to federal, state, or tocal laws

SECTION VIII SPECIAL PROTECTION INFORMATION

ion Protection particulate respirator is desirable. under extreme dusting conditions a standard

Ventilation Local Exhaust Recommended 1 Special

Protective Gloves

Other Protective

Eye Protection

₹

Smock, apron, goggles, proper gloves, eye wash station Chemical safety glasses.

SPECIAL PRECAUTIONS

Precautions to be Taken SECTION IX

in Handling and Storing keep container lightly closed when not in use Store in a cool, dry place away from acid or acid fumes Wash thoroughly after handling.

Read label on container before using

Other Precautions

(B)

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Rev. No. For laboratory use only. Not for drug, tood or household use. Keep out of reach of children 16. 1 Date 1/12/87 Approved Maintaill - Lective Coordinator

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